Equations						
Sum of the results						
$\overline{No. of observations}$						
$Max - Min \qquad \qquad Max - Min \qquad \qquad$						
$\Delta R = \frac{1}{n}$						
Random Uncertainty						
$\underline{Percentage \ Uncertainty} = \underline{\underline{Mean}} \times 100\%$						
Key Words	Meaning					
Uncertainty	How close the calculated result may be to the actual value. All experiments have					
	uncertainties					
Error	A word often used in place of uncertainty in some Journals					
Accuracy	Results which are accurate have a mean value very close to the real value.					
Des sisis a	Accurate results may be spread over a wide range.					
Precision	Results which are precise will be spread over a very small range. This does not					
	mean they will be close to the real value. A result can be precise but not					
SI Units	dccurate! The International System of Units (SI) states the 7 base units that must be used in					
	all calculations. These are: Second (time). Meter (distance), Kilogram (mass).					
	Ampere (current), Kelvin (temperature), Mole (amount of a substance), and					
	Candela (light intensity). All other units are derived from these 7.					
	https://www.npl.co.uk/si-units					
Mean	Mean is the type of average used in physics. A mean value of serval results should					
	always be found when analysing experiments.					
Random	Random uncertainties are present in every experiment. This is the reason we					
	MUST repeat all measurements.					
Scale Reading	The scale reading is the uncertainty that accords because of the scale on the					
Digital	device used. These are never completely precise.					
Digital	the uncertainty in a digital scale (such as an electronic balance) can be taken as					
Analoguo	$\pm 1$ in the smallest significant figure displayed. (my scales can read $\pm 0.2$ kg)					
Analogue	The uncertainty in an analogue scale (such as an metre stick) can be taken as $\pm 1/2$					
	in the smallest scale division.					
Systematic	A systematic effect is a type of error that skews all results by the same amount in					
	the same direction. E.g. slow running clocks, scales not set to zero					
Zero Error	This is the most common systematic error. It occurs because the zero point of the					
	device is not set perfectly. E.g. the end of a meter stick has worn away. This can					
	be eliminated by starting all measurements from a different value on the metre					
	stick.					
Reducing	We can attempt to reduce the random uncertainty by taking multiple observations					
Uncertainty	and calculating a mean from these observations. The more measurements the					
Percentage	Detice: Dercentage uncertainty is when you convert an absolute uncertainty to a $\%$ by					
reiteillage	dividing it by the reading or mean: "Value Units $\pm$ percentage uncertainty $\%$ "					
Overall	The largest percentage uncertainty is the best estimate of the overall uncertainty $\sqrt{2}$					
uncertainty	in a experiment A result including this should be in the form. "Value Units +					
and the tailing	percentage uncertainty %"					
Absolute	This is the 'total' uncertainty in an experiment. A result including this should be in					
uncertainty	the form: "(Value ± absolute uncertainty) units"					
	······································					

## ACCURACY V PRECISION

Both accuracy and precision reflect how close a measurement is to an actual value, but they are not the same. Accuracy reflects how close a measurement is to a known or accepted value, while precision reflects how reproducible measurements are, even if they are far from the accepted value. Measurements that are both precise and accurate are repeatable and very close to true values.

SIGNIFICANT FIGURES - NOT THE SAME AS DECIMAL PLACES AND CAN BE SLIGHTLY DIFFERENT TO MATHS

<ul> <li>Your answer should have a consistent number of significant figures number of significant figures the least significant data given in the question. E.g if a question give you data at 3 significant figures then you should give your answer to significant figure, however through out the questions</li> <li>DO NOT write down all the numbers you see on your calculator.</li> <li>DO use the FIX or SCI button if you feel confident about using them.</li> <li>Odo planning and experimentation normally avoids this mismatch values and significant figures.</li> <li>Good planning and experimentation normally avoids this mismatch values and significant figures.</li> <li>Try not to choose apparatus where you can obtain too few or an unnecessarily high number of significant figures.</li> </ul>	<ul> <li>about the significant</li> <li>quote a value</li> <li>rite down all rs you see on ator.</li> <li>FIX or SCI ou feel about using</li> <li>Tri ur</li> <li>Tri fig</li> </ul>	ly in a calculation , only the last line! ve a consistent number of significant figures as ta given in the question. E.g if a question gives at figures then you should give your answer to 3 ever through out the questions e final answer. I in the exam if you use too many significant erimentation normally avoids this mismatch of figures. ratus where you can obtain too few or an aber of significant figures. I rements to the same number of significant
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Prefix	<u>Symbol</u>	<u>Multiple</u>	Multiple in full	Quantity	Symbol	Unit & Unit Symbol
Tera	Т	x10 <sup>12</sup>	x1 000 000 000 000	Mass	m	kilogram kg
Giga	G	x10 <sup>9</sup>	x1 000 000 000	i i i i i i i i i i i i i i i i i i i		intogram, ng
Mega	Μ	x10 <sup>6</sup>	x1 000 000	Length	1	metre, m
Kilo	K	x10 <sup>3</sup>	x1 000	Time	t	second, s
Centi	С	x10 <sup>-2</sup>	÷100	Temperature	Т	degrees Celsius,
Milli	Μ	x10 <sup>-3</sup>	÷1 000			Kelvin, K
Micro	μ	x10 <sup>-6</sup>	÷1 000 000	Current	1	ampere, A
Nano	N	x10 <sup>-9</sup>	÷1 000 000 000			I]
Pico	Р	x10 <sup>-12</sup>	÷1 000 000 000 000			

#### REDUCING UNCERTAINTIES

- Always try to get the most accurate piece of apparatus to do the job.
- Analogue scales can be read to the nearest 0.5 of a scale division.
- Repeat measurements to reduce the random uncertainty in an experiment. You must repeat measurements in your Assignments

SCALE READING UNCERTAINTY

This value indicates how well an instrument scale can be read.

An estimate of reading uncertainty for an **analogue scale** is generally taken as:

 $\pm$  half the least division of the scale.

For a **digital scale** it is taken as

± 1 in the least significant digit displayed.



The greatest percentage uncertainty in any one measurement is taken as the overall unicertainty in the experiment. When comparing uncertainties, it is important to take the percentage in each. This percentage uncertainty is often a good estimate of the percentage uncertainty in the final numerical result of the experiment.

eg if one measurement has an uncertainty of 3% and another has an uncertainty of 5%, then the overall percentage uncertainty in this experiment should be taken as 5%

# QUANTIFYING UNCERTAINTIES



### Which experiment has the best design?



B, not offset, average is the true value, little spread of readings (ie giving a small approximate random uncertainty)

C, beginners luck, now evidence that this is the real value as you've no repeats

D, small random uncertainty but a systematic effect. Might show in a graph, where best fit line doesn't pass through the origin.

Example

A student sets up the apparatus in the diagram to measure the average acceleration of a model car as it travels between P and Q



#### SQA Higher Paper 1994

During an experiment to measure the specific heat capacity of a liquid the relationship  $VIt = mc\Delta T$  is used

The following quantities are measured

 $V= 12.0 \pm 0.1 V$   $I = 4.2 \pm 0.1 A$   $t = 300 \pm 1s$   $m=500\pm 2 g$  $DT = 15 \pm 1 \ ^{\circ}C$   $V = \frac{0.1}{12} = 0.8\%$   $I = \frac{0.1}{4.2} = 2\%$   $t = \frac{1}{300} = 0.3\%$   $m = \frac{2}{500} = 0.4\%$  $DT = \frac{1}{15} = 7\%$  Which quantity will contribute the largest uncertainty to the final answer for the specific heat capacity, c? A Voltage B Current

- C Time
- D mass
- E temperature difference

# OPEN ENDED QUESTIONS

Any number of answers may be given which are equally correct (or incorrect

0 marks.- not demonstrated any understanding of the physics.

1 mark.- demonstrates a limited understanding of the physics.

2 marks.- demonstrates a reasonable understanding of the physics.

**3 marks**. -demonstrates a good understanding of the physics. Might include a statement of the principles involved, a relationship and the application.

This does not mean the answer has to be what might be termed an "excellent" answer or a complete one.

### Strategy for Solving Open-Ended Questions.

In answering open ended questions you should:

- 1) read the question taking care not to skim read.
- 2) reread the question.
- 3) try to understand/define the problem situation and what is asked.
- 4) visualise the situation.
- 5) draw a diagram and include any relevant information such as speeds, velocities, forces, vector directions etc.
- 6) determine and write down
- a) relevant physics principles e.g. conservation of momentum
- b) note area/topic of physics involved in problem e.g. internal resistance of supplies
- c) relationship relevant to variables in the problem.
- 7) use knowledge of familiar quantities such as body mass, body height, length of running track to create estimated values as required.
- 8) with the information noted previously solve the problem or do what can be done.
- 9) reread the response to see if it makes sense and answers the question.